

A Summary of MAGE: A Method for Estimating the Maximum Possible Chemical Energy Content of UCG Product Gas per Unit Area for a Multistrata Coal Zone

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# A Summary of MAGE: A Method for Estimating the Maximum Possible Chemical Energy Content of UCG Product Gas per Unit Area for a Multistrata Coal Zone

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# **Auspices Statement**

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Underground coal gasification (UCG) has the potential to extract energy from unmineable coals in a form of syngas with no CO<sub>2</sub> emissions. The selection of potential sites for UCG, however, is a complicated process that involves consideration of various scientific, technological, and economical issues.

The present report describes a new method for estimating the upper bound on the chemical energy content of UCG product gas per square meter of land area. The product gas would have about this much chemical energy content if there were no water influx, no roof collapse or spalling above the coal zone, no heat lost to the surroundings, no gas losses to the surroundings, and 100% conversion of the coal to produced gas products. This method can particularly be used for comparison of different regions and selection of locations that potentially may provide the highest amounts of energy in the form of UCG product gas. Note that other factors such as geomechanical, physical, and hydrological properties of coal and rock, seam dip, proximity of aquifers, surface infrastructure (e.g., mines, cities, power plants, pipelines), land cost, etc., should also be considered in the site selection process.

It is commonplace in the coal industry to arrive at a summed energy content of multiple coals seams by summing over the seams of interest the product of the as-received coal heating value, times an estimate of the coal density, times the thickness of the seam to produce a total heating value on an areal basis (MJ per m² of land area). The method described in this report more accurately fits the energy balance for underground coal gasification in that it takes into account the energy required to heat the coal ash, coal moisture, parting minerals, partings moisture, and product gas to their final temperature in a UCG operation.

We call this value MAGE, for  $\underline{M}$  aximum  $\underline{A}$  real  $\underline{G}$  as chemical  $\underline{E}$  nergy in the product gas from a zone containing one or more coal seams.

The estimates for a specific coal basin are based on the information obtained from numerous boreholes. For each borehole, we usually know the depths of all minerals and coals, i.e., in fact we know the thicknesses of all the layers under the surface along the borehole. The coalcontaining layers are typically characterized using proximate analysis, which provides weight percentages of moisture, ash, volatiles, and fixed carbon (the sum of these components is equal to 100 %). Each such layer is also characterized by the lower heating value (*LHV*), which constitutes the combustion enthalpy of the material at 25°C minus the latent enthalpy of water vaporization. Often, the layer density is also known.

The lower heating value of the coal  $(LHV_c)$  is often used for comparison of mined coals. In the case of UCG, however, it is more appropriate to use the lower heating value of the product gas  $(LHV_g)$ . If carbon reacts with oxygen and steam with no energy loss, LHV of the obtained gas mixture is equal to that of carbon. This is explained by the conservation of energy (for example, the energy obtained from hydrogen combustion compensates for the energy that needs to be spent for hydrogen formation). The UCG process involves some inherent energy losses that have to be subtracted from  $LHV_c$  in the estimates of  $LHV_g$ . Three major mechanisms of energy losses in the coal layer during UCG, accounted for in the present estimate, are:

- Vaporization of water contained in the layer and heating the obtained steam to the product gas temperature ( $T_a$ , defined later)
- Heating ash or any other solid remained in the layer to the rubble temperature ( $T_r$ , defined later)
- Heating the dry gas products to  $T_q$

An additional energy loss occurs due to heat transfer to the rock layers located between coal layers. As noted above, heat loss to the surroundings is neglected. Often, however, the parallel coal layers are separated by relatively thin layers of rock. The UCG reactor size may exceed the thickness of a single coal layer if the separating rock layer collapses during the cavity growth. In this case, the rock layer will consume the energy for:

- ullet Vaporization of water contained in the layer and heating the obtained steam to  $T_g$
- Heating rock to  $T_r$

In the estimates, such thin layers of rock can be treated as the coal layers with zero  $LHV_c$  and no gas products (except for steam). Thus the formula for calculating  $LHV_g$  of the UCG product gas from a single coal or rock layer is:

$$LHV_g = LHV_c - (h_{st} - h_w) \cdot mf_w - c_{pm} \cdot (T_r - T_o) \cdot mf_m - c_{pg} \cdot (T_g - T_o)$$

$$\cdot mr_g$$
(1)

where  $h_{st}$  is the enthalpy of steam at  $T_g$ ,  $h_w$  is the enthalpy of liquid water at  $T_o$ ,  $c_{p\,m}$  is the average specific heat of mineral (ash or rock),  $c_{p\,g}$  is the average specific heat of product gas,  $T_o$  is the initial temperature of coal and rock,  $mf_w$  is the mass fraction of water in the layer,  $mf_m$  is the mass fraction of mineral (ash or rock) in the layer, and  $mr_g$  is the mass ratio of the dry product gas to the coal. The latter parameter is determined by the formula:

$$mr_g = mf_c \cdot (1 - mf_w - mf_m) \cdot \frac{M_g}{M_c}$$
(2)

where  $mf_c$  is the mass fraction of carbon in dry ash-free coal,  $M_g$  is the molar mass of the product gas, and  $M_c$  is the molar mass of carbon. Note that for a rock layer, Eq. 1 produces a negative value, which means that the layer does not generate any chemical energy (product gas) and only consumes energy.

The maximum areal gas energy of the layer is determined by the formula:

$$AE = \rho \cdot \delta \cdot LHV_g \tag{3}$$

where  $\rho$  is the density of coal or rock in the layer and  $\delta$  is the thickness of the layer. Note that the dimension of AE is the unit energy per unit area, e.g.,  $MJ/m^2$ .

In Eq. 3, the thickness,  $\delta$ , is the *vertical* thickness, i.e., the difference between the bottom depth and the top depth of the layer along the borehole. Here we neglect a potential effect of coal seam dip (which is often unknown). For high dip angles, more careful considerations will be needed. During calculations, the operator selects a series of adjacent layers that could be considered as a single UCG reactor. It is recommended that the bottom and top layers consist of coal and the interior rock layers are sufficiently thin to be collapsed during UCG. The maximum areal gas energy of the potential UCG reactor is determined as the sum of AE values of all the layers from the top to the bottom:

$$AE_R = \sum_{i=1}^n AE_i \tag{4}$$

The parameters  $mf_w$  and  $mf_m$ , and  $LHV_c$  are typically known for each coal-containing layer. For rock layers,  $LHV_c = 0$ , while the values of 0.15 and 0.85 can be assumed for  $mf_w$  and  $mf_m$ , respectively. The parameter  $mf_c$  has also to be known. If it is unknown, typical content of carbon in dry ash-free coal for the specific type of coal (e.g., lignite) can be used.

The product gas temperature,  $T_g$ , is defined as the temperature of gas at the bottom of the production well, which can be considered as the exit of the UCG reactor. It is assumed that the energy required for heating gases (including steam) to higher temperatures, that take place in some places inside the UCG reactor, is not lost and is stored as part of a chemical energy of the product gas. Based on previous UCG experience, it is recommended to assume that both  $T_g$  and  $T_r$  are equal to 500°C. The initial temperature of coal or rock,  $T_o$ , is equal to 25°C.

The estimate requires knowledge of the product gas composition, which in general is dependent on the location (e.g., increasing the depth leads to increasing methane content) and injected gas (air vs. oxygen). For simplicity, however, it is recommended to assume some specific gas composition, which is expected for most locations if the same injected gas (e.g., oxygen) is used, for example, 35% H<sub>2</sub>, 11% CH<sub>4</sub>, 11% CO, and 43% CO<sub>2</sub>.

The enthalpies of steam  $h_{st}$  and liquid water  $h_w$  are determined from steam tables. The specific heat of the product gas,  $c_{pg}$ , is determined by averaging the specific heats of the individual gas components at  $T_o$  and  $T_g$  and applying the formula for the specific heat of a gas mixture. It is assumed that the specific heat of mineral  $c_{nm} = 1 \text{ kJ/kg/K}$ .

If the layer density is unknown, reasonable assumptions can be made (e.g., the density of lignite is 1400 kg/m<sup>3</sup> and the density of clay is 1800 kg/m<sup>3</sup>). Note that the density for the actual (wet) material should be used.